

Communication

Air-assisted High-performance Field-effect Transistor with Thin Films of Picene

Hideki Okamoto, Naoko Kawasaki, Yumiko Kaji, Yoshihiro Kubozono, Akihiko Fujiwara, and Minoru Yamaji

J. Am. Chem. Soc., 2008, 130 (32), 10470-10471 • DOI: 10.1021/ja803291a • Publication Date (Web): 16 July 2008

Downloaded from http://pubs.acs.org on February 8, 2009



More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

View the Full Text HTML





Air-assisted High-performance Field-effect Transistor with Thin Films of Picene

Hideki Okamoto,*^{,†} Naoko Kawasaki,[‡] Yumiko Kaji,[‡] Yoshihiro Kubozono,^{*,‡} Akihiko Fujiwara,[§] and Minoru Yamaji¹

Division of Chemistry and Biochemistry, Okayama University, Okayama 700-8530, Japan, Research Laboratory for Surface Science, Okayama University, Okayama 700-8530, Japan, Japan Advanced Institute of Science and Technology, Ishikawa 923-1292, Japan, and Department of Chemistry and Chemical Biology, Gunma University, Kiryu 376-8515, Japan

Received May 3, 2008; E-mail: hokamoto@cc.okayama-u.ac.jp; kubozono@cc.okayama-u.ac.jp

Field-effect transistors (FETs) with thin films of organic molecules have many advantages such as large-area coverage, mechanical flexibility, lightweight, and a low-temperature/low-cost fabrication process. The field-effect mobility, μ , value has been still lower than the values for inorganic MOS FETs,¹ but that of FET with single crystals of organic molecules reached up to 40 cm² V⁻¹ s^{-1.2} The μ values of FETs with thin films of some organic molecules are also comparable to those of amorphous Si FETs, approximately 1 cm² V⁻¹ $s^{-1.3}$ Currently, pentacene molecule is generally used for active layer of p-channel thin film FET because it can realize a high μ value of approximately 1 cm² V⁻¹ s⁻¹.⁴ Nevertheless, instability or degradation of pentacene under atmospheric conditions causes severe problems for FET performance. Therefore, post-pentacene molecules, materials more suitable as active layer of FETs, have been searched.⁵ The aromatic hydrocarbon of picene possesses the same number of benzene rings as pentacene (Figure 1a), and the expanded and delocalized π -system exists within the molecule. Furthermore, picene should be more chemically stable than pentacene because of the large energy band gap of picene (3.3 eV) and large ionization potential (5.5 eV) (Supporting Information). This may guarantee a stable FET operation under atmospheric conditions. Herein, we report the first FET device with thin films of "old" polycyclic aromatic hydrocarbon, picene. Excellent p-channel enhancement-type FET characteristics with a μ value of 1.1 cm² V⁻¹ s⁻¹ and an on-off ratio of 1.6×10^5 have been observed under O2 atmosphere.

The top-contact device structure of picene thin film FET is shown in Figure 1a. The picene sample was synthesized by our group according to a new synthesis method (Supporting Information), and the ¹H NMR spectrum shows no impurities (Figure S1 in Supporting Information). The details of device fabrication and dimension of device are described in Supporting Information.

The fabricated device was transferred from evaporation chamber to measurement vessel; that is, the device was exposed to air for more than 30 min. First, the FET performance at room temperature was measured under vacuum of 10^{-6} Torr without any annealing, and the temperature dependence of FET performance was investigated from 230 to 295 K under vacuum. Second, stability of device performance was investigated for a long time, more than 20 days, at 300 K. Third, after the device was once exposed to air for 1 h, the FET performance was again investigated under vacuum of 10^{-6} Torr and under atmospheric conditions. The surface topological image and photoemission spectrum were measured with an atomic force microscope (AFM: Seiko Instruments SPA-400) and a photoemission spectrometer (Riken AC-2).

Panels b and c of Figure 1 show p-channel enhancement-type output and transfer curves ($V_{DS} = -100$ V) of picene FET at 290 K. The μ



Figure 1. (a) Device structure of picene thin film FET. (b) Output and (c) transfer curves under vacuum, and (d) transfer curve after exposure to 160 Torr of O_2 for 4 h. In (b), V_G is varied at -10 V step from 0 to -100 V.

was determined to be 0.11 cm² V⁻¹ s⁻¹ from the transfer curve in forward measurement mode (forward transfer curve; Figure 1c), while the μ value estimated from the transfer curve in reverse measurement mode (reverse transfer curve; Figure 1c) was 0.27 cm² V⁻¹ s⁻¹; the average value of μ values estimated from the forward transfer curves for three devices are described in Supporting Information so as to show a precision of the determined μ value. The on-off ratio was approximately 10⁴. This FET showed very high threshold voltages, V_{TH} , of 67 V (forward mode) and 80 V (reverse mode).

As shown in Figure 2a, the μ increases with an increase in temperature in the low temperature region under vacuum of 10^{-6} Torr, indicating a thermally activated hopping like transport for the channel conduction. However, the μ decreases with an increase in temperature above 275 K. This result may imply a different transport mechanism, such as a Drude-type conduction, which is found for organic single crystal FETs and thin film FETs with high μ value.^{6,7} The intrinsic mobility, μ_0 , activation energy, ε_a , and the ratio of total density of states (DOS) for trap states, $N_{\rm t}$, to the total DOS of valence band (highest occupied molecular orbital (HOMO)), N_v , were determined to be 0.13 cm² V⁻¹ s⁻¹, 180 meV, and 4.6 \times 10⁻⁵, respectively, from the $\mu - T$ plot for the forward transfer curve using a formula based on multiple trap and release (MTR) model.⁸ Here, the ε_a corresponds to the energy difference between the energy, ε_t , of trap states and that, ε_v (or ε_{HOMO}), of valence band (i.e., $\varepsilon_a = \varepsilon_t - \varepsilon_v$). The μ_0 is relatively high, and the ε_a is close to those for organic FETs reported so far (100-200 meV).⁸⁻¹⁰ Furthermore, the N_t/N_v is extremely small in comparison with those reported so far for thin film FET devices $(10^{-1}-10^{-2})$ ¹⁰ and the value is close to that for single crystal FET (4×10^{-6}) ⁹ The above values for μ_0 , ε_a , and N_t/N_v are almost the same as those estimated from the $\mu - T$ plot for the reverse transfer

[†] Division of Chemistry and Biochemistry, Okayama University.

^{*} Research Laboratory for Surface Science, Okayama University

[§] Japan Advanced Institute of Science and Technology.

^{II} Gunma University.



Figure 2. (a) μ -*T* and (b) μ -*t* of picene thin film FET under vacuum. (c) AFM and (d) energy diagram of picene thin films. Fitting lines in (a) and (b) are drawn on the basis of MTR model and exponential decay, respectively.

curve ($\mu_0 = 0.16 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, $\varepsilon_a = 130 \text{ meV}$, and $N_t/N_v = 6.3 \times$ 10^{-4}). These results indicate that the total DOS of trap states is very small under vacuum.

Time dependence of μ at 300 K for picene thin film FET under vacuum is shown in Figure 2b. The μ decreases exponentially with an increase in t, and it dropped down to approximately $0.02 \text{ cm}^2 \text{ V}^{-1}$ s^{-1} at t = 480 h. The FET device was exposed to air for 1 h after being kept under vacuum of 10⁻⁶ Torr for 480 h, and then the FET characteristics were measured under vacuum. The μ recovered from 1.6×10^{-2} to 0.13 cm² V⁻¹ s⁻¹ in the forward transfer curve and from 1.9 \times 10 $^{-2}$ to 0.47 cm^2 V^{-1} s^{-1} in the reverse transfer curve. The recovery of μ was also confirmed in the measurement under atmospheric conditions. The other FET devices with picene thin films recorded high μ values of 1.05 cm² V⁻¹ s⁻¹ in the reverse transfer curve after air exposure for around 1000 h (Figure S2 in Supporting Information) and $1.1 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ in the reverse transfer curve after O_2 exposure for 4 h (Figure 1d). The on–off ratio of $1.6\,\times\,10^5$ is recorded in the forward transfer curve of the picene FET device exposed to O_2 for 4 h (Figure 1d). The very high μ value of 1.75 cm² V^{-1} s⁻¹ is also observed in the reverse transfer curve of picene FET exposed to O₂ for 70 h (Figure S3). Typical output characteristics after O₂ exposure are shown in Figure S4. Thus, air (or O₂) exposure to FET produced a remarkable increase in both values of μ and on-off ratio. This is contrast to that in the other FETs where air exposure produced an increase in μ and a decrease in on-off ratio.¹¹ This implies that air (or O_2) really assists the channel transport in picene FET. Furthermore, it presents possible applications toward O2 or H2O sensor of picene FET.

The hysteresis in the transfer curve becomes larger under air than that under vacuum, as shown in Figure 1c and Figure S2 in Supporting Information. The air exposure of organic FET may enhance biasstressed trap states, as suggested by pentacene FET.¹² This effect can explain well the enhancement of hysteresis under atmospheric conditions. Therefore, the small N_t/N_y found from the $\mu-T$ plots (Figure 2a) may be confined to the interface between active layer and gate dielectric under vacuum; that is, the larger N_t/N_y may be observed under atmospheric conditions. The pure O2 exposure did not show a clear enhancement of hysteresis (Figure 1d and Figures S3 and S4). This fact suggests that the origin of hysteresis is not O2 but H2O in air.

When an FET device is exposed to air or O₂, holes should be doped into the trap states at the interface between picene and SiO₂ gate dielectric because O₂ (or O) and H₂O (or OH) are electron acceptors. This can produce small ε_a because of hole filling into trap states, leading to high effective field-effect mobility. In addition, charge neutrality between doped acceptor ions $(O_2^- \text{ or } OH^-)$ and filled holes may reduce scattering of carriers in the channel, which would produce high μ value. It is interesting to identify which the channel transport is assisted by, O₂ or H₂O. Judging from the increases in μ and on-off ratio found for pure O₂-exposed picene FET, the channel transport in picene FET may be assisted by O₂ (Supporting Information).

The AFM image of picene thin film on SiO₂/Si substrate shows a dense packing of large grains (approximately 500 nm; Figure 2c). These grains show a very flat surface within the roughness of approximately 1 nm. Thus, the topological image suggests that the picene molecules are well ordered in the grains, and that thin films of picene are preferable for formation of π -conduction network. The energy diagram of picene (Figure 2d) shows energy difference of 0.6 eV between the Fermi energy, $\varepsilon_{\rm F}$, of Au and $\varepsilon_{\rm HOMO}$ of picene, while that between the $\varepsilon_{\rm F}$ and the energy, $\varepsilon_{\rm LUMO}$, of lowest unoccupied molecular orbital (LUMO) is very large (approximately 2.7 eV); the $\epsilon_{\rm F}$ and $\epsilon_{\rm HOMO}$ are determined from the onset of photoemission from picene thin film. The ε_{HOMO} is consistent with that estimated by DFT calculation with basis set of B3LYP/6-31G. The ε_{LUMO} is determined by consideration of the HOMO-LUMO gap, 3.3 eV, determined from the onset of absorption spectrum (not shown). The small energy difference between $\varepsilon_{\rm F}$ and $\varepsilon_{\rm HOMO}$ can explain well p-channel conduction in picene FET with Au electrodes and lack of n-channel conduction.

In conclusion, high performance p-channel FET with μ more than $1 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ has been realized with picene thin films, and the FET properties are clearly improved under air/O2 conditions. This shows that the picene FET is an air (or O₂)-assisted device.

Acknowledgment. This work was supported by a Grant-in-Aid (18340104) and Kyoto-Advanced Nanotechnology Network of the MEXT, Japan.

Supporting Information Available: Details of stability and synthesis of picene, ¹H NMR, device fabrication, transfer/output curves of FET device, and O₂ exposure effect. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- Dimitrakopoulos, C. D.; Malenfant, P. R. L. Adv. Mater. 2002, 14, 99.
 Takeya, J.; Yamaguchi, M.; Tominari, Y.; Hirahara, R.; Nakazawa, Y.; Nishikawa, T.; Kawase, T.; Shimoda, T.; Ogawa, S. Appl. Phys. Lett. 2007, 90, 102120.
- Zaumseil, J.; Sirringhaus, H. Chem. Rev. 2007, 107, 1296.
- (4) Lin, Y.-Y.; Gundlach, D. J.; Nelson, S. F.; Jackson, T. N. IEEE Electron Device Lett. 1997, 18, 606.
- (a) Meng, H.; Bao, Z.; Lovinger, A. J.; Wang, B.-C.; Mujsce, A. M. J. Am. (a) Weilg, H., Bab, Z., Wolfger, A. J., Wallg, D. C., Majse, A. M. and Chem. Soc. 2001, 123, 9214. (b) Merlo, J. A.; Newman, C. R.; Gerlach, C. P.; Kelley, T. W.; Muyres, D. V.; Fritz, S. E.; Toney, M. F.; Frisbie, C. D. J. Am. Chem. Soc. 2005, 127, 3997. (c) Meng, H.; Sun, F.; Goldfinger, M. B.; Gao, F.; Londono, D. J.; Marshal, W. J.; Blackman, G. S.; Dobbs, W. F.; Statum, S. C. 2005, 127, 3097. (c) Meng, H.; Sun, F.; Goldfinger, M. B.; Gao, F.; Londono, D. J.; Marshal, W. J.; Blackman, G. S.; Dobbs, K. D.; Keys, D. E. J. Am. Chem. Soc. 2006, 128, 9304. (d) Ponomarenko, S. A.; Kirchmeyer, S.; Elschner, A.; Alpatova, N. M.; Halik, M.; Klauk, H.; Zschieschang, U.; Schmid, G. Chem. Mater. 2006, 18, 579. (e) Locklin, J.; Ling, M. M.; Sung, A.; Roberts, M. E.; Bao, Z. Adv. Mater. 2006, 18, 2989. (f) Klauk, H.; Zchieschang, U.; Weitz, R. T.; Meng, H.; Sun, F.; Nunes, G.; Keys, D. E.; Fincher, C. R.; Xiang, Z. Adv. Mater. 2007, 19, 3882. (g) Koppe, M.; Scharber, M.; Brabec, C.; Duffy, W.; Heeney, M; McCulloch, I. Adv. Funct. Mater. 2007, 17, 1371
- (6) Podzorov, V.; Menard, E.; Borissov, A.; Kiryukhin, V.; Rogers, J. A.; Gershenson, M. E. Phys. Rev. Lett. 2004, 93, 086602
- Nelson, S. F.; Lin, Y.-Y.; Gundlach, D. J.; Jackson, T. N. Appl. Phys. Lett. 1998, 72, 1854.
- Knipp, D.; Street, R. A.; Völkel, A. R. *Appl. Phys. Lett.* **2003**, *82*, 3907. Tripathi, A. K.; Heinrich, M.; Siegrist, T.; Pflaum, J. Adv. Mater. **2007**, (9)
- 19.2097 (10) Kawasaki, N.; Ohta, Y.; Kubozono, Y.; Konishi, A.; Fujiwara, A. Appl. Phys. Lett. 2008, 92, 163307.
- Meijer, E. J.; Detcheverry, C.; Baesjou, P. J.; van Veenendaal, E.; de Leeuw, D. M.; Klapwijk, T. M. J. Appl. Phys. **2003**, *93*, 4831.
- (12)Lang, D. V.; Chi, X.; Siegrist, T.; Sergent, A. M.; Ramirez, A. P. Phys. Rev. Lett. 2003, 93, 086802.

JA803291A